

Kenai Peninsula Borough School District
Science: Chemistry
Unit 3: PERIODIC TABLE - USES

NGSS Standards:

HS-PS1-1. Use the periodic table as a model to predict the relative properties of elements based on the patterns of electrons in the outermost energy level of atoms.

HS-PS1-2. Construct and revise an explanation for the outcome of a simple chemical reaction based on the outermost electron states of atoms, trends in the periodic table, and knowledge of the patterns of chemical properties.

HS-PS1-3. Plan and conduct an investigation to gather evidence to compare the structure of substances at the bulk scale to infer the strength of electrical forces between particles.

HS-PS1-7. Use mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction.

HS-PS1-8. Develop models to illustrate the changes in the composition of the nucleus of the atom and the energy released during the processes of fission, fusion, and radioactive decay.

HS-PS2-6. Communicate scientific and technical information about why the molecular-level structure is important in the functioning of designed materials.

HS-PS3-2. Develop and use models to illustrate that energy at the macroscopic scale can be accounted for as a combination of energy associated with the motions of particles (objects) and energy associated with the relative positions of particles (objects).

MATH STANDARDS:

MP.2

Reason abstractly and quantitatively. (HS-PS3-1),(HS-PS3-2),(HS-PS3-3),(HS-PS3-4),(HS-PS3-5) HS-PS1-2),(HS-PS1-3),(HS-PS1-7),(HS-PS1-8)

- a. decontextualize to abstract a given situation and represent it symbolically and manipulate the representing symbols.
- b. reflect during the manipulation process in order to probe into the meanings for the symbols involved
- c. create a coherent representation of the problem
- d. make sense of quantities and their relationships in problem situations
- e. attend to the meanings of quantities
- f. use flexibility with different properties of operations and objects
- g. translate an algebraic problem to a real-world context
- h. explain the relationship between the symbolic abstraction and the context of the problem

- i. compute using different properties
- j. consider the quantitative values, including units, for the numbers in a problem

MP.4	<p>Model with mathematics. (HS-PS3-1),(HS-PS3-2),(HS-PS3-3),(HS-PS3-4),(HS-PS3-5) (HS-PS1-2),(HS-PS1-3),(HS-PS1-7),(HS-PS1-8)</p> <p>In grades 9-12 mathematically proficient students will:</p> <ul style="list-style-type: none"> • apply mathematics to solve problems in everyday life, society, and workplace • identify important quantities in a practical situation and map the relationships using such tools as diagrams, two-way tables, graphs, flowcharts and formulas • consistently interpret mathematical results in the context of the situation and reflect on whether the results make sense • apply knowledge, making assumptions and approximations to simplify a complicated situation, realizing that these may need revision later • make assumptions and approximations to simplify a situation, realizing the final solution will need to be revised • identify important quantities in a practical situation and map their relationships using such tools as diagrams, two-way tables, graphs, and formulas • analyze quantitative relationships to draw conclusions • improve the model if it has not served its purpose
HSN.Q.1	<p>Use units as a way to understand problems and to guide the solution of multi-step problems; choose and interpret units consistently in formulas; choose and interpret the scale and the origin in graphs and data displays. (HS-PS2-1),(HS-PS2-2),(HS-PS2-4),(HS-PS2-5),(HS-PS2-6) (HS-PS1-2),(HS-PS1-3),(HS-PS1-7),(HS-PS1-8)</p>
HSN.Q.2	<p>Define appropriate quantities for the purpose of descriptive modeling. (HS-PS2-1),(HS-PS2-2),(HS-PS2-4),(HS-PS2-5),(HS-PS2-6) (HS-PS.7),(HS-PS1-8)</p>
HSN.Q.3	<p>Choose a level of accuracy appropriate to limitations on measurement when reporting quantities. (HS-PS2-1),(HS-PS2-2),(HS-PS2-4),(HS-PS2-5),(HS-PS2-6) (HS-PS1-2),(HS-PS1-3),(HS-PS1-7),(HS-PS1-8)</p>

ELA Standards:

RST.11-12.1	<p>Cite specific textual evidence to support analysis of science and technical texts, attending to important distinctions the author makes and to any gaps or inconsistencies in the account. (HS-PS2-1),(HS-PS2-6)</p>
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RST.9-10.7	Translate quantitative or technical information expressed in words in a text into visual form (e.g., a table or chart) and translate information expressed visually or mathematically (e.g., in an equation) into words. (HS-PS1-1)
WHST.9-10.2	Write informative/explanatory texts, including the narration of historical events, scientific procedures/ experiments, or technical processes. (HS-PS2-6)
WHST.11-12.8	8. Gather relevant information from multiple authoritative print and digital sources, using advanced searches effectively; assess the usefulness of each source in answering the research question; integrate information into the text selectively to maintain the flow of ideas, avoiding plagiarism and following a standard format for citation. (HS-PS1-3)
WHST.9-12.9	Draw evidence from informational texts to support analysis, reflection, and research. (HS-PS1-3)
SL.9-12.5	Make strategic use of digital media (e.g., textual, graphical, audio, visual, and interactive elements) in presentations to enhance understanding of findings, reasoning, and evidence and to add interest. (HS-PS3-1),(HS-PS3-2),(HS-PS3-5)

ESSENTIAL QUESTIONS

1. Explain the significance and rows and columns in the periodic table.
2. Describe the relationship between atom and element?

BIG IDEAS:

1. Understand how to use the periodic table.
2. Know the different sections of the periodic table.

Vocabulary: Group, Family, Period, Metal, Nonmetal, Noble gas, Metalloids, Halogen, Alkali metals, Alkali earth metals, Octet rule, Periodic trend, Symbol